

Toxic Metals (Pb, Cd, Ni and As) in Tubers grown around a Lead-zinc Derelict Mine and their significance to Health and Phytoremediation

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ABSTRACT

Proton Induced X-ray Emission (PIXE) technique was used to analyze toxic metals in the leaves (aerial part) and roots (underground part) of four common tubers namely *Colocasia and Xanthosoma* (Cocoa yam), *Manihol eaculenhis* (Cassava), *Dioscoreu rolundata* (White Yam) and *Ipomoea batatas* (Sweet potato) grown around Enyigba lead-zinc derelict. The results revealed that heavy metals in soil (mean pH = 6.5 ± 0.29) decreased in the order Pb > Cd > Ni > As. The mean concentration (mg kg^{-1}) of heavy metals in soil was found in the range of Pb (0.01-0.41); As (0.02-0.54); Cd (0.02-0.12); and Ni (8.24-34.66). The results revealed that levels of Pb in white yam (0.41 mg/kg) and cassava (0.32 mg/kg); As in cassava (0.54 mg/kg), white yam (0.12 mg/kg) and sweet potato (0.14 mg/kg) and Cd in cocoyam (0.12 mg/kg) all exceeded the World health Organisation Maximum Limit (WHO ML) and thus they are unfit for human consumption. High values of Translocation Factor (TF) observed in sweet potatoes for cadmium (2.0) and in cocoyam for cadmium (1.2) suggest that the affected tubers were potential hyperaccumulators. The variation in the parameters determined were found to be statistically significant ($p < 0.05$) as determined by one way analysis of variance.

Keywords: Heavy Metals, Tubers, Bioaccumulation factor, Translocation factor, PIXE, Enyigba, lead-zinc derelict

INTRODUCTION

Tubers are fleshy underground food-storing plant. It is a fleshy swollen part of a root such as a dahlia root or of an underground stem such as a potato that stores food over winter and produces new growth in spring. In Nigeria like most African countries, tubers form the major staple for most families. Young plants developing from tuber buds are nourished from starch stored in the tubers until mature enough to develop root systems. Heavy metals from the environment enter into plants via either the roots and/or foliar surfaces [1]. Metal uptake and accumulations are affected by some factors such as soil pH, metal solubility, conductivity, stages of plant growth, plant species, soil type and fertilizers [2]. Different plants have different capacity to absorb and accumulate heavy metals which often leads to contamination of the food chain and as a result causes varying degrees of illness based on acute and chronic exposures. There are three major bases for classifying a plant as a hyperaccumulator for heavy metal(s). First is when Translocation factor (TF) is greater than one ($TF > 1$). Translation Factor is shoot/root quotient (the ratio of concentration of heavy metal in the shoot to the root). Second is when Bioaccumulation factor (BAF) is greater than one ($BAF > 1$). Bioaccumulation factor is the extraction coefficient which is the level of heavy metal in the shoot divided by total level of heavy metal in the soil. Extraction coefficient gives the proportion of total heavy metal in the soil which is taken up by the plant shoot/aerial part of the plant [3]. Third is when there is higher level of heavy metal of 10 - 500 times the levels in normal plants (uncontaminated plants). The percentage of

heavy metals accumulated by plant varies according to the heavy metals involved and previous studies have shown plants accumulating more than 1000 mg/kg of dry weight for nickel, copper, cobalt, chromium or lead and more than 10,000 mg/kg for zinc or manganese [4]. Capacity for accumulation is as a result of hypertolerance, (*phytotolerance*) which came from adaptative evolution of plants in hostile environments along multiple generations [5].

The present study was carried out within the vicinity of a lead-zinc derelict mine, Enyigba located 14 Km South of Abakaliki (latitudes 4°20' and 7°00'N and longitudes 5°25' and 9°35'). Enyigba is a rural area with sparse rural population of farmers who cultivate their crops especially tubers around the mine waste. More importantly, the community serve as food basket of Abakaliki city as farmers regularly supply their farm produce to the town. Mothers often fry or boil potatoes and white yam which is served as breakfast to young school children and adults. Cocoyam is usually boiled overnight and served as breakfast at the family table. Cassava is commonly prepared by fermenting it into "fufu" or further drying and frying to give what is known as "garri" and it they are served as a daily staple all over Nigeria. The contamination status of many edible plants especially tubers from Enyigba is yet to be established. Previous works in Enyigba [6,7] focused on soils and few plants respectively. However, this present study focused on two aims. The first one was to determine the Pollution Index or the contamination status of the soil where the tubers are planted. The second one was to use bioaccumulation factor (BAF) and translocation factor (TF) to evaluate the metal uptake abilities of the investigated tubers which gives a clue to their phytoremediation potentials [8].

METHODOLOGY

Collection of Soil Samples: Composite top-soil and sub-soil samples were collected, 20 m apart at 0-30 cm (n=6) and at 60-90 cm (n=6) depths respectively. The former represents the top soil while the later represents the sub soil [9]. The soil samples were air-dried, ground mechanically with stainless steel soil grinder and sieved to obtain < 2 mm fraction. 30 g sub-sample was drawn from the bulk (< 2 mm fraction) and reground with laboratory mortar and pestle to obtain < 200 µm fraction. The sample was further dried in an open inert vessel in a muffle furnace at 105 °C for 2 hours so as to remove soil moisture, after which the samples were cooled in desiccators [10].

Collection of Tubers Samples: Samples of *Colocasia* and *Xanthosoma* (Cocoa yam), *Manihot eaculenhis* (Cassava), *Dioscoreu rolundata* (White Yam) and *Ipomoea batatas* (Sweet potato) were collected, authenticated, labelled and stored in pre-treated storage containers. For this work the fleshy underground part is regarded as the root. The roots and leaves of the tubers were separated in each case and the components were cut into pieces. The plant tissues were cleaned to remove dust, soil and other particles by putting them through a three step washing sequence [10]. First they were washed with water, then with P-free detergent and followed by de-ionized water. The moisture and water droplets were removed with the help of blotting papers. The samples were air dried, and placed in a dehydrator at approximately 80°C for 48 hours. This was followed by mechanical grinding with the aid of an agate mortar which was done with utmost care to avoid contamination with the elements to be analyzed. The ground tissues were further dried at 65°C in an oven to obtain a constant weight upon which to base the analysis [11].

PIXE Analyses of Heavy Metal in Plant and Soil Samples: The process begins with pelletization of the samples using CAVER model manual palletizing machine at a pressure of 6 - 8 torr to give 13mm pellets. Proton Induced X-ray Emission (PIXE) spectrometer component of 1.7 MV NEC model Tandem Nuclear Accelerator was used for the metal determination. Dried plant and soil samples were analyzed for heavy metal contents using a procedure similar to one described by [13] This was done by irradiating the target sample in PIXE spectrometer for 3 minutes in a vacuum chamber with 3 MeV protons (beam currents of 10-70 nA and beam diameter of 4 mm). Bombardment with ions of sufficient energy (MeV protons) produced by an ion accelerator, will cause inner shell ionization of element atoms in the sample. Outer shell electrons drop down to

replace inner shell vacancies and only certain transitions are allowed. X-rays of a characteristic energy of the element are emitted. An energy dispersive detector is used to record and measure these X-rays [14]. The X-rays generated from the target were measured with two Si/Li detectors and the corresponding metal concentration was ascertained by means of incorporated computer device.

In addition to concentration of metals in the soil samples, soil pH; percentages of sand, silt and clay as well as percentages of organic matter were determined using Orion 920A pH meter [15,17], Hydrometer method [18] and Walkley and Black method [16] respectively

Statistical Analysis: Composite samples of the investigated soils and plants were assayed and analyzed individually in triplicates. Data generated from PIXE were reported as mean + Standard Deviation. One way analysis of variance (ANOVA) and Fisher's Least Square Difference (LSD) were used to determine significant difference within and between groups, considering a level of significance of less than 5% ($p < 0.05$) by using Minitab 2007 version 13.6 statistical software [19]. PI, BAF and TF were calculated from the generated data.

RESULTS

Tables 1 and 2 present the mean concentrations of heavy metals and the properties of the soils of Enyigba derelict respectively. Table 3 presents the mean concentration of heavy metals in the investigated tubers while Tables 4 and 5 showed their corresponding translocation and bioaccumulation factors respectively.

Table 1: PIXE Mean Concentrations (mg/kg) of Heavy Metals in Soil of Enyigba Derelict and their Pollution Indices

Metal	Enyigba (n = 3)				
	Topsoil	PI	Subsoil	PI	US-EPA
As	4.8 ± 1.8	0.06	2.12 ± 1.6	0.03	75*
Cd	126.0 ± 42	1.5	28.8 ± 6.2	0.34	85
Ni	82.6 ± 22.0	1.1	34.8 ± 8.2	0.46	75
Pb	1116.8 ± 43.2	2.7	91.7 ± 16.7	0.22	420

* Values refer to metal concentration in typical soils [20], PI = Pollution index.±

Table 2: Properties of Soil from Enyigba Mine

Properties	Enyigba Mine (n = 3)
Sand (%)	61.28 ± 5.2
Silt (%)	7.12 ± 0.8
Clay (%)	31.60 ± 2.6
Organic Matter (%)	1.34 ± 0.5
pH	6.5 ± 0.29

Table 3: Concentrations of Heavy Metals in Common Tubers in Enyigba Derelict (n=3)

Heavy metals concentration (mg/kg)						
Botanical name	Common name	Plant parts	Pb	As	Cd	Ni
<i>Colocosia</i>	& Cocoyam	Leaves	0.01	ND	0.12	12.24
<i>Xanthosama</i>		Root	0.22	0.02	0.10	22.82
<i>Manihot esculentus</i>	Cassava	Leaves	0.12	0.38	0.10	34.66
		Root	0.32	0.54	0.10	34.02
<i>Dioscorea rotundata</i>	White yam	Leaves	0.21	ND	ND	12.42
		Root	0.41	0.12	0.02	14.86
<i>Ipomoea batatas</i>	Sweet potato	Leaves	0.04	0.02	0.04	8.24
		Root	0.12	0.14	0.02	10.24
WHO/ FAO	Maximum	Limit	0.3	0.1	0.1	67

Table 4: Translocation Factor of Heavy Metals of Common Tubers from Enyigba Derelict

Botanical name	Common name	Pb	As	Cd	Ni
<i>Colocosia and Xanthosama</i>	Cocoyam	0.05	-	1.2	0.54
<i>Manihot esculentus</i>	Cassava	0.38	0.70	1.0	1.01
<i>Dioscorea rotundata</i>	White yam	0.51	-	-	0.84
<i>Ipomoea batatas</i>	Sweet potato	0.33	0.14	2.0	0.81

Table 5: Bioaccumulation Factor of Heavy Metals of Common Tubers from Enyigba Derelict

Botanical name	Common name	Pb	As	Cd	Ni
<i>Colocosia and Xanthosama</i>	Cocoyam	0.000	0.003	0.001	0.299
<i>Manihot esculentus</i>	Cassava	0.000	0.132	0.001	0.585
<i>Dioscorea rotundata</i>	White yam	0.000	0.017	0.000	0.231
<i>Ipomoea batatas</i>	Sweet potato	0.000	0.023	0.001	0.158

DISCUSSION

Heavy metals in Soil

From Tables 1 and 2, the mean concentrations of the heavy metals in the soil samples were observed to decrease with depth. It is a known fact that when concentration of pollutants are significantly higher in top soil than sub soil, it suggests that the source of the pollutants is of anthropogenic origin [15]. The obtained results revealed that heavy metal concentration in the Enyigba derelict mine decrease in the order Pb > Cd > Ni > As in topsoil and Pb > Ni > Cd > As > Cr in the subsoil but only Pb, was significant at p < 0.05. These results are at variance both in trend and values obtained for each metal with the earlier results of previous work [7,21]. The difference is due to the sensitivity of PIXE technique used in this present study compared to AAS technique used by other researchers in the previous work. In addition, environmental dependent parameters such as metal concentrations and pH, are never static due to changes in temperature, volume and nature of human perturbation on the natural ecosystem. Three of the heavy metals namely Pb, Ni and Cd exceeded the US-EPA Regulatory Limits [22] and their pollution indices were in the order, Pb (2.7) > Cd (1.5) > Ni (1.1). The overall pollution status of soils was 1.96 which confirms that Enyigba soils are already polluted with heavy metals [7,21]. Soils with PI > 1 are known to affect plants, animal and ultimately distort the food chain they support [15].

The percentage of organic matter content of Enyigba soil was 1.34 % while the soil pH was 6.5. The extent of soil pollution with heavy metals and subsequent uptake by crops depend on organic matter content among other factors. Organic matter and pH values have been reported to independently and associatively influence the concentrations of heavy metals in soils. Soils with pH around 7.0 have higher availability of nutrient elements such as Mg, Ca, K, N and S, while metals such as Pb is less soluble and therefore less available at about this pH [23]. The soil parameters results were comparable to another similar research on lead in soil and vegetations [24]. On the basis of organic matter and pH, the results however indicate that the soils from the study areas can support agricultural activities efficiently in agreement with similar findings [21,25,26]. However on the basis of heavy metal indices, Enyigba soil was already polluted.

Heavy metal in Tubers

Lead: The concentration of Pb in the roots of white yam (0.41 mg/kg) and cassava (0.32 mg/kg) exceeded the WHO maximum limit (Table 3). Lead is the main cause for concern in this work because it is highly toxic at minute concentration and can be harmful to man who may consume the effected tubers. Presence of lead affects the gastrointestinal tract, kidneys, and central nervous system. Children exposed to lead are at risk for impaired development, lower IQ, shortened attention span, hyperactivity, and mental deterioration while , while adults usually experience decreased reaction time, loss of memory, nausea, insomnia, anorexia, and weakness of the joints when exposed to lead (ATSDR, 1993). In many plants, Pb accumulation can exceed several hundred times the threshold of WHO ML (Wong, 2002). Tubers usually show ability to accumulate large amounts of lead without visible changes in their appearance or yield. In this work, Pb was observed to have accumulated more in the roots than in the leaves unlike the result of Muhammad *et al.*, (2008) where higher concentration of Pb was observed in the leaves of studied plants than in other roots. The reason may have to do with the difference in pH of the soils and the percentage of organic matter which are known to promote Pb uptake [27].

Arsenic: Arsenic was detected in all the investigated tubers except in the leaves of cocoyam and white yam from Enyigba mine (Tables 3). The concentration of As exceeded WHO maximum Limit (WHO ML) in white yam and sweet potato. Low BAF and TF of As were observed in all the studied plants (Tables 4 and 5). This suggests that though As level was high in white yam and potato but they are ineffectiveness in accumulating As from the soil to the aerial part of the plants. Immobilizations As and other metals depend on soil pH, amount of organic matter, and texture Soil. It is a known fact [28, 29, 30] that arsenate [As(V)] absorbed to soil at pH 4-7 while arsenite [As(III)] at pH 7-10. But since speciation study was not carried out in this work, pH may be responsible for the low uptake of As by the tubers [31,32].

Cadmium: Cadmium accumulated more in the leaves than in the root of the some of the tubers (Table 3). Cadmium in cassava (0.01 mg/kg) is at critical point which means that the concentration was of the same value as WHO ML but it exceeded in cocoyam (0.12 mg/kg). Environmentally speaking cassava can be effectively used for biomonitoring of Cd [33,34]. Although Low BAF values of Cd were observed generally, however high TF of Cd observed were in the order: sweet potato (2.0) > cocoyam (1.2) > cassava (1.0) (Tables 4 and 5). A key trait of metal hyperaccumulators is the efficient metal transport from roots to shoots, characterized by the TF being greater than one [35]. Based on the foregoing, the TF >1 observed in sweet potato and cocoyam means they are potential Cd-hyperaccumulators [31,32].

Nickel: Nickel concentrations in all the studied tubers were below WHO ML (Table 3). This suggests that all the studied tubers were free of Ni contamination and therefore are safe for human consumption. Nickel at this level is not a known toxic metal to human health [36,37]. Excess and deficiency of Ni in tubers are detrimental to human health [38]. Deficiency of Nickel have been linked with hyperglycemia, hypertension, depression, sinus congestion, fatigue, reproductive failures and growth problems in humans, while excess intake of Ni leads to hypoglycemia, asthma, nausea, headache,

and epidemiological symptoms like cancer of nasal cavity and lungs. The prescribed safety limit of Nickel is 3 to 7 mg/day in humans. Low BAF and TF were observed except TF of Zn in cassava (1.13) which is greater than one (Tables 4 and 5). Cassava based on $TF > 1$, is a potential Ni-hyperaccumulator [39].

CONCLUSION

From this present work, Concentrations of Cd in cocoyam; Pb and As, in Cassava and in white yam in addition to As sweet potato all exceeded the WHO ML and therefore are unfit for human consumption. Their simple pollution indices (the ratio of their concentrations to WHO ML) were all above one. To avoid food crisis, edible plants such as the studied tubers are not considered best to be employed as phytoremediation agents. From this study, and on the basis of $TF > 1$, cocoyam and sweet potato hyperaccumulated of Cd just as cassava is a potential hyperaccumulator of Cd, and Ni. The implication of all these is that while consumption of these plants may likely result to metal toxicity in man, these plants can be useful in cleaning up the contaminated mine derelict.

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REFERENCES

1. Sawidis, T., Chettri, M. K., Papaionnou, A., Zachariadis, G., & Stratis. J. (2001). A study of Metal Distribution from Lignite Fuels Using Trees as Biological Monitors. *Ecotox. Environ. Safe*, 48, 27-35. <http://dx.doi.org/10.1006/eesa.2000.2001>
2. Sharma. R., Agrawal, M., and Marshall, F. (2006). Heavy Metals Contamination in Vegetables Grown in Wastewater Irrigated Areas of Varanasi. *India Bulletin of Environmental Contamination and Toxicology*, (77), 312-318.
3. Reeves, R.D. (2003). Tropical Hyperaccumulators of Metals and their Potential for Phytoextraction, *Plant and Soil*. (249), 57-65.
4. Vassil, A.D., Kapulnik, Y., Raskin, I., and Salt, D.E. (1998). The Role of EDTA in Lead Transport and Accumulation by Indian Mustard, *Plant Physiology*, (117), 447-491.
5. Boyd, R.S. and Martens, S.N. (2003). The Significance of Metal Hyperaccumulation for Biotic Interactions, *Phytoremediation Journal*, 2(16), 123-125
6. Nweke, F. N, Okaka, N.C and Offor, E. C (2008). Lead, zinc and pH concentrations of Enyigba soil in Abakaliki Local Government Area of Ebonyi State. Nigeria. *African Journal of Biotechnology* 7(14), 2441-2443
7. Chukwuma, S.C. (1994), Evaluating Baseline Data for Lead (Pb) and Cadmium (Cd) in Rice, Yam, Cassava, and Guinea Grass from Cultivated Soils in Nigeria, *Toxicological and Environmental Chemistry*, Vol. 45, pp. 45-56.
8. Mattina, M.J., Berger W.L; Musante C. and White I. C. (2006). Concurrent plant uptake of heavy metals and persistent organic pollutants from soil. *Environmental Pollution* 124: 375-378

9. Danish Environmental Protection Agency (2000). Guidance Regarding Advice of Inhabitants of Slightly Contaminated Soil (in Danish). Danish EPA, Office of Soil Contamination: Copenhagen, (75) pp. 132-138 (On-line: www.mst.dk/udgiv/publikationer/2000/87-7944-302-8/pdf)
10. Reuter, D.J., Robinson, J.B., Peverill, K.I. and Price, G.H. (1988) Guidelines for Collecting, Handling and Analyzing Plant Material. *In*: Reuter, D.J. and Robinson, J.B (eds) Plant Analysis an Interpretation Manual. Inkarta Press, Melbourne pp 20-30.
11. Jones, J.B., Wolf, B. and Mills, H.A (1991). Plant Analysis Handbook, Micro-Macro Publishing, Inc., Athens, GA., pp 213
12. Food and Agriculture Organization (FAO) (2009). Soil Testing and Plant Analysis, Bullentin. No 38/1, Food and Agriculture Organization, Rome, Italy, pp 7-12
13. Olabanji, S. O, Buoso, M. C, Ceccato, D.; Haque, A. M. I, Cherubini, R., and Moschini, G. (1995). *Nucl. Instr. And Meth. in Phys. Res.*, 405-407.
14. Cahill, T.A., (1981). *Nucl. Sci. Particle Induced X-ray Emission*. ASM Metals Handbook, 9th Ed.. Vol. 10:213.
15. Alloway BJ and Ayres DC (1997). Chemical principles for environmental pollution Blackie Academic and Professional pp 190-220.
16. White, R. (2006). Principles and Practice of Soil Science, Blackwell Publishers, Wiley and Sons Inc, New York, pp 140-241
17. Klute F. (1986) Methods of Soil Analysis and Physical mineralogical Method. *American Society of Agronomy*, Madison, WI, USA 23-24
18. Brown, M.T. and Hall, I.R. (1990). Ecophysiology of Metal Uptake by Tolerant Plants, *In*: Shaw AJ (ed.) Heavy Metal Tolerance in Plants: Evolutionary Aspects. Boca Raton: FL: CRC Press, pp 95-104.
19. Minitab Tip Sheet 8 (2007) Department of Mathematics and Statistics University of Reading, (On-line: <http://http://www.reading.ac.uk/statistics/tipsheets>)
20. Miroslav, R. and Vladimir, N. B (1999). Practical Environmental Analysis. Royal Society of Chemistry, UK, pp 284, 314 and 357.
21. Nweke, F. N., Okaka, A. N. C., & Offor, E. C. (2008). Lead, zinc and pH concentrations of Enyigba soil in Abakaliki Local Government Area of Ebonyi State, Nigeria. *African journal of Biotechnology*, 7(14), 2441-2443.
22. USEPA. (1993). Cleaning up the Nation's Waste Sites; Markets and Technology Trends, United States Environmental Protection Agency, (On-line: <http://www.epa.gov/phytoem/0243.html>)

23. Assuncao A., Bookum W., Nelissen H., Vooijs R., Schat H. and Ernst W. (2003). Differential Metal- Specific Tolerance and Accumulation Patterns among *Thlaspu caerulescens* Populations Originating from Different Soil Types. *New Phytoremediation Journal* (159) 411-419.
24. Nwoko, C.O. and Egunobi, J.K., (2002). Lead Contamination of Soil and Vegetation in an Abandoned Battery Factory Site in Ibadan, *Nigerian Journal of Sustained Agricultural Environment*, 4(1), 91 - 96
25. Anikwe, M.A.N. and Nwobodo, K.C.A.. (2002). Long Term Effect of Municipal Waste Disposal on Soil Properties and Productivity of Sites Used for Urban Agriculture in Abakaliki Nigeria. *Bioresources Technology* (83) 241- 251.
26. Nnabo, P.N., Orazulike D.M. and Offor, O.C. (2011) The Preliminary Assessment of The Level of Heavy Elements Contaminations in Stream Bed Sediments of Enyigba and Environs, South Eastern *Nigeria Journal of Basic Physical Research* Vol. 2, No.2, pp 43 - 52,
27. Demirezen, D. and Ahmet, A. (2006). Heavy Metal Levels in Vegetables in Turkey. *Journal of Food Quality* (29), 252-265.
28. McBride, M.B. (1994). Environmental Chemistry of Soils, Oxford University Press, pp 42-50.
29. Ma, L.Q., Komar. K.M., Tu, C., Zhang, W., Cai, Y. and Kenelly, E.D. (2001) A Fern that Hyperaccumulate Arsenic. *Nature* (409) 579-582
30. Zhang WH, Cai Y, Tu C, Ma QL (2002) Arsenic speciation and distribution in an arsenic hyperaccumulating plant. *Sci Environ* 300: 167-177
31. Reeves, R. and Baker, A. (2000). Metal-Accumulating Plants, John Wiley and Sons Inc, pp 193-229.
32. Fayiga AQ, Ma LQ (2006) Using Phosphate Rock to Immobilize, Metals in Soils and Increase Arsenic Uptake in *Pteris vittata*. *Sci Total Environ* 359: 17-25
33. Frieberg, L., Ellinder, G. and Kjellstrom, T. (1992). Cadmium - Environmental Aspects: World Health Organisation, *Environmental Health Criteria* Geneva. (134).1123-1128.
34. Zhao FJ, Jiang RF, Dunham SJ, McGrath SP (2006) Cadmium uptake, translocation and tolerance in the Hyperaccumulator *Arabidopsis halleri*. *New Phytolerance Journal* (172), 646-654
35. Baker, A.J. and Brooks, R.R. (1989) Terrestrials Higher Plants which Hyper Accumulate Metallic Elements. A Review of their Distribution, Ecology and Phytochemistry, *Biorecovery Journal* 4(1), 81-26.
36. Joseph, W.G. (1991). Plants for Toxicity Assessment. (ASTM) Committee E-47 on Biological Effects and Environmental Fate. Joy Press, pp 112-133
37. MacNicol, R.D and Beckett, P.H.T. (1985). Critical Tissue Concentrations of Potentially Toxic Elements. *Plant Soil* (85),107-129
38. Oti W.J.O. (2015) Pollution Indices and Bioaccumulation Factors of Heavy Metals in Selected Fruits and Vegetables from a Derelict Mine and their Associated Health Implications. *Int J. Environ Sci. Toxic. Res.* Vol. 3(1) pp 9-15

39. Wong S.C., Li, D.X. and Zhang, G. (2002) Environmental Pollutants, Board Press, China pp 258-266